

3-(1-Methylpiperidinio)-1-propanesulfonate

**Katarzyna D. Koclega,^{a,b}
Maksymilian Chruszcz^a and
Wladek Minor^{a*}**

^aDepartment of Molecular Physiology and Biological Physics, University of Virginia, 1300 Jefferson Park Avenue, Charlottesville, VA 22908, USA, and ^bTechnical University of Lodz, Faculty of Biotechnology and Food Sciences, Institute of Technical Biochemistry, Stefanowskiego 4/10, 93-923 Lodz, Poland

Correspondence e-mail:
wladek@iwonka.med.virginia.edu

Key indicators

Single-crystal X-ray study

$T = 106\text{ K}$

Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$

R factor = 0.032

wR factor = 0.085

Data-to-parameter ratio = 10.5

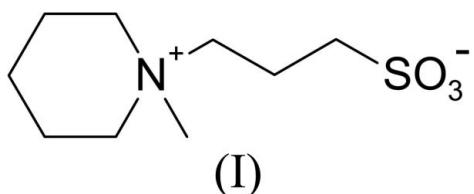
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title sulfobetaine (NDSB-221), $\text{C}_9\text{H}_{19}\text{NO}_3\text{S}$, is used for protein solubilization and stabilization. The piperidine ring has a chair conformation, with the sulfopropyl group in an equatorial position. The trimethylene spacer adopts a fully extended conformation.

Received 5 December 2006
Accepted 7 December 2006

Comment

3-(1-Methylpiperidinio)-1-propanesulfonate (NDSB-221), (I), belongs to a group of non-detergent sulfobetaines (NDSBs) used as solubilization and renaturation agents in protein purification (Vuillard *et al.*, 1995; Goldberg *et al.*, 1995). Other compounds from this group are used in isoelectric focusing, differential scanning calorimetry (Collins *et al.*, 2006) and protein crystallization (Vuillard *et al.*, 1994; Vuillard *et al.*, 1996).



The most important interactions for the structural stability of (I) are between charged quaternary ammonium and sulfonate groups. There are several short contacts (Table 1) in which H and O atom distances are at least 0.3 Å shorter than the sum of the van der Waals radii.

It has been suggested (Vuillard *et al.*, 1995) that the conformation of the sulfopropyl group in NDSBs may be important for interactions with proteins during the solubilization process. In the crystal structure of (I) reported in this paper, the torsion angles S1—C1—C2—C3 and C1—C2—C3—N1 have values of 174.22 (11) and 179.28 (13)°, respectively, and thus the three-methylene linker has an extended conformation, which is similar to what was observed in the cases of trimethylammoniopropane sulfonate (Yokoyama *et al.*, 2003) and 3-(ethyldimethylammonio)propane sulfonate (Koclega *et al.*, 2006). The piperidine ring of (I) has a chair conformation, with the sulfopropyl group in an equatorial position.

Experimental

NDSB-221, (I), was purchased from Anatrace. Crystallization was performed at room temperature and the crystal used for the X-ray diffraction experiment was obtained by slow evaporation of a propan-2-ol solution.

Crystal data

$C_9H_{19}NO_3S$
 $M_r = 221.31$
Monoclinic, $P\bar{1}/c$
 $a = 8.300 (1) \text{ \AA}$
 $b = 9.485 (1) \text{ \AA}$
 $c = 13.993 (1) \text{ \AA}$
 $\beta = 96.396 (4)^\circ$
 $V = 1094.75 (19) \text{ \AA}^3$

$Z = 4$
 $D_x = 1.343 \text{ Mg m}^{-3}$
Cu $K\alpha$ radiation
 $\mu = 2.51 \text{ mm}^{-1}$
 $T = 106 (2) \text{ K}$
Plate, colourless
 $0.52 \times 0.13 \times 0.05 \text{ mm}$

Data collection

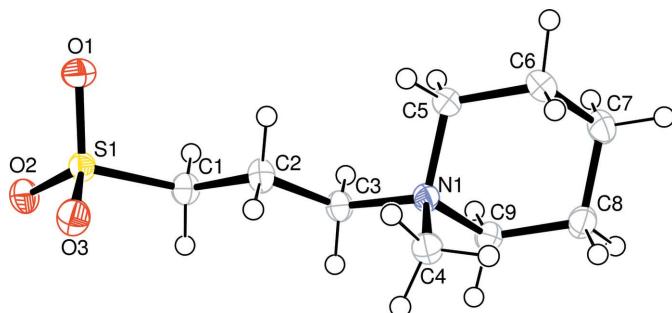
Rigaku R-AXIS RAPID diffractometer
 ω scan with χ offset
Absorption correction: multi-scan (Otwinowski *et al.*, 2003)
 $T_{\min} = 0.69$, $T_{\max} = 0.88$

4068 measured reflections
2140 independent reflections
2063 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.018$
 $\theta_{\text{max}} = 72.3^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.032$
 $wR(F^2) = 0.085$
 $S = 1.10$
2140 reflections
204 parameters
All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0376P)^2 + 0.6719P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.23 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.31 \text{ e \AA}^{-3}$
Extinction correction: SHELXL97 (Sheldrick, 1997)
Extinction coefficient: 0.0034 (3)

**Figure 1**

The molecular structure of NDSB-221, (I). Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as spheres of arbitrary radii.

3000SM, ORTEPIII (Burnett & Johnson, 1996) and ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: HKL-3000SM.

The authors thank Matthew D. Zimmerman for helpful discussion. This work was supported by contract GI11496 from HKL Research Inc.

References

- Burnett, M. N. & Johnson, C. K. (1996). ORTEPIII. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.
- Collins, T., D'Amico, S., Georlette, D., Marx, J. C., Huston, A. L. & Feller, G. (2006). *Anal. Biochem.* **352**, 299–301.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Goldberg, M. E., Expert-Bezançon, N., Vuillard, L. & Rabilloud, T. (1995). *Fold. Des.* **1**, 21–27.
- Koclega, K. D., Chruszcz, M. & Minor, W. (2006). *Acta Cryst.* **E62**, o5757–o5759.
- Minor, W., Cymborowski, M., Otwinowski, Z. & Chruszcz, M. (2006). *Acta Cryst.* **D62**, 859–866.
- Otwinowski, Z., Borek, D., Majewski, W. & Minor, W. (2003). *Acta Cryst.* **A59**, 228–234.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Vuillard, L., Baalbaki, B., Lehmann, M., Nørager, S., Legrand, P. & Roth, M. (1996). *J. Cryst. Growth*, **168**, 150–154.
- Vuillard, L., Braun-Breton, C. & Rabilloud, T. (1995). *Biochem. J.* **305**, 337–343.
- Vuillard, L., Rabilloud, T., Leberman, R., Berthet-Colominas, C. & Cusack, S. (1994). *FEBS Lett.* **353**, 294–296.
- Yokoyama, T., Murakami, G., Akashi, H. & Zenki, M. (2003). *Anal. Sci.* **19**, 805–806.

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C4}-\text{H4A}\cdots\text{O1}^{\text{i}}$	0.973 (19)	2.329 (19)	3.2409 (19)	155.7 (15)
$\text{C4}-\text{H4C}\cdots\text{O2}^{\text{ii}}$	0.973 (18)	2.414 (18)	3.364 (2)	165.0 (15)
$\text{C4}-\text{H4B}\cdots\text{O3}^{\text{iii}}$	0.95 (2)	2.46 (2)	3.3753 (19)	161.3 (15)
$\text{C8}-\text{H8A}\cdots\text{O3}^{\text{iii}}$	0.99 (2)	2.40 (2)	3.390 (2)	173.9 (16)
$\text{C3}-\text{H3B}\cdots\text{O2}^{\text{iv}}$	0.977 (19)	2.494 (19)	3.3949 (19)	153.2 (15)
$\text{C9}-\text{H9B}\cdots\text{O2}^{\text{iv}}$	0.966 (18)	2.342 (19)	3.285 (2)	165.1 (14)

Symmetry codes: (i) $-x, -y, -z + 1$; (ii) $-x + 1, -y, -z + 1$; (iii) $x, -y - \frac{1}{2}, z + \frac{1}{2}$; (iv) $x, -y + \frac{1}{2}, z + \frac{1}{2}$.

All H atoms were located in electron-density difference maps and their positional and displacement parameters were refined freely [$\text{C}-\text{H} = 0.95 (2)-1.03 (19) \text{ \AA}$].

Data collection: HKL-2000 (Otwinowski & Minor, 1997); cell refinement: HKL-2000; data reduction: HKL-2000; program(s) used to solve structure: HKL-3000SM (Minor *et al.*, 2006) and SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: HKL-3000SM and SHELXL97 (Sheldrick, 1997); molecular graphics: HKL-